

## REB, The Course

### Class 7 Practice Assignment Calculations

Equation numbers correspond to the Class 7 Practice Assignment Solution.

Rates of the mechanistic steps:

$$r_j = k_{j,f} \prod_{i_r} [i_r]^{-\nu_{i_r,j}} - k_{j,r} \prod_{i_p} [i_p]^{\nu_{i_p,j}}$$

$$r_2 = k_{2,f} [E] [C] - k_{2,r} [E-C] \quad (5)$$

$$r_3 = k_{3,f} [E-C] [S] - k_{3,r} [S-E-C] \quad (6)$$

$$r_4 = k_{4,f} [S-E-C] \quad (7)$$

Rate of the apparent reaction:

$$r_{i,j_{non}} = \sum_{j'} \nu_{i,j'} r_{j'}$$

$$r_{P,1} = r_4 \quad (8)$$

Bodenstein steady-state approximations:

$$0 = \sum_{j'} \nu_{i_{RI},j'} r_{j'}$$

$$0 = -r_2 \quad (9)$$

$$0 = r_2 - r_3 + r_4 \quad (10)$$

$$0 = r_3 - r_4 \quad (11)$$

Catalyst conservation equation:

$$C_{cat,0} = C_{cat,free} + \sum_{i_c} \kappa_{i_c} C_{i_c}$$

$$E_0 = [E] + [E-C] + [S-E-C] \quad (12)$$

Most abundant intermediate inequalities:

$$[E-C] \gg [E] \quad (13)$$

$$[E-C] \gg [S-E-C] \quad (14)$$

Substitute (5) in (9) and rearrange.

$$[E-C] = \frac{k_{2,f}[C]}{k_{2,r}} [E] \quad (15)$$

Substitute (6) and (7) in (11) and rearrange, then substitute (15).

$$0 = k_{3,f}[E-C][S] - k_{3,r}[S-E-C] - k_{4,f}[S-E-C]$$

$$[S-E-C] = \frac{k_{3,f}[E-C][S]}{k_{3,r} + k_{4,f}}$$

$$[S-E-C] = \frac{k_{2,f}k_{3,f}}{k_{2,r}(k_{3,r} + k_{4,f})} [S][C][E] \quad (16)$$

Substitute (15) and (16) in (12) and solve for [E].

$$E_0 = [E] + \frac{k_{2,f}[C]}{k_{2,r}} [E] + \frac{k_{2,f}k_{3,f}}{k_{2,r}(k_{3,r} + k_{4,f})} [S][C][E]$$

$$[E] = \frac{E_0}{\left( 1 + \frac{k_{2,f}}{k_{2,r}} [C] + \frac{k_{2,f}k_{3,f}}{k_{2,r}(k_{3,r} + k_{4,f})} [C][S] \right)} \quad (17)$$

Substitute (17) into (15) and (16) to get expressions for [E-C] and [S-E-C].

$$[E-C] = \frac{\frac{k_{2,f}}{k_{2,r}} E_0 [C]}{1 + \frac{k_{2,f}}{k_{2,r}} [C] + \frac{k_{2,f} k_{3,f}}{k_{2,r} (k_{3,r} + k_{4,f})} [C] [S]} \quad (18)$$

$$[S-E-C] = \frac{\frac{k_{2,f} k_{3,f}}{k_{2,r} (k_{3,r} + k_{4,f})} E_0 [C] [S]}{\left( 1 + \frac{k_{2,f}}{k_{2,r}} [C] + \frac{k_{2,f} k_{3,f}}{k_{2,r} (k_{3,r} + k_{4,f})} [C] [S] \right)} \quad (19)$$

Substitute (7) in (8) and then substitute (19) in the resulting equation.

$$r_{P,1} = k_{4,f} [S-E-C]$$

$$r_{P,1} = \frac{\frac{k_{2,f} k_{3,f} k_{4,f}}{k_{2,r} (k_{3,r} + k_{4,f})} E_0 [C] [S]}{\left( 1 + \frac{k_{2,f}}{k_{2,r}} [C] + \frac{k_{2,f} k_{3,f}}{k_{2,r} (k_{3,r} + k_{4,f})} [C] [S] \right)} \quad (20)$$

Substitute (17) and (18) in (14) and simplify.

$$\frac{\frac{k_{2,f}}{k_{2,r}} E_0 [C]}{1 + \frac{k_{2,f}}{k_{2,r}} [C] + \frac{k_{2,f} k_{3,f}}{k_{2,r} (k_{3,r} + k_{4,f})} [C] [S]} \gg \frac{E_0}{\left( 1 + \frac{k_{2,f}}{k_{2,r}} [C] + \frac{k_{2,f} k_{3,f}}{k_{2,r} (k_{3,r} + k_{4,f})} [C] [S] \right)}$$

$$\frac{k_{2,f}}{k_{2,r}} [C] \gg 1$$

Delete the 1 in the denominator or (20).

$$r_{P,1} = \frac{\frac{k_{2,f}k_{3,f}k_{4,f}}{k_{2,r}(k_{3,r} + k_{4,f})} E_0 [C] [S]}{\left( \frac{k_{2,f}}{k_{2,r}} [C] + \frac{k_{2,f}k_{3,f}}{k_{2,r}(k_{3,r} + k_{4,f})} [C] [S] \right)} \quad (21)$$

Substitute (17) and (19) in (14) and simplify.

$$\frac{\frac{k_{2,f}}{k_{2,r}} E_0 [C]}{1 + \frac{k_{2,f}}{k_{2,r}} [C] + \frac{k_{2,f}k_{3,f}}{k_{2,r}(k_{3,r} + k_{4,f})} [C] [S]} \gg \frac{\frac{k_{2,f}k_{3,f}}{k_{2,r}(k_{3,r} + k_{4,f})} E_0 [C] [S]}{\left( 1 + \frac{k_{2,f}}{k_{2,r}} [C] + \frac{k_{2,f}k_{3,f}}{k_{2,r}(k_{3,r} + k_{4,f})} [C] [S] \right)}$$

$$\frac{k_{2,f}}{k_{2,r}} [C] \gg \frac{k_{2,f}k_{3,f}}{k_{2,r}(k_{3,r} + k_{4,f})} [C] [S]$$

Drop the term on the right from the denominator in (21) and simplify.

$$r_{P,1} = \frac{\frac{k_{2,f}k_{3,f}k_{4,f}}{k_{2,r}(k_{3,r} + k_{4,f})} E_0 [C] [S]}{\left( \frac{k_{2,f}}{k_{2,r}} [C] \right)}$$

$$r_{P,1} = \frac{k_{3,f}k_{4,f}}{(k_{3,r} + k_{4,f})} E_0 [S] \quad (22)$$